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# Antioxidant capacity and phytochemical characterization of *Spathodea campanulata* growing in different climatic zones in Brazil

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# ABSTRACT

The aim of present study was evaluating the antioxidant activity and real content of phytochemicals of *Spathodea campanulata* from different climatic regions (two subtropical and one tropical area/cities) at Brazil. Samples of leaves, flowers and nectars from *S. campanulata* were tested for total phenolic content and *in vitro* antioxidant activity by ORAC (Oxygen Radical Absorbance Capacity), FRAP (Ferric Reducing Antioxidant Power) and DPPH (2,2-diphenyl-1-picrylhydrazyl) radical scavenging assay. The phytochemical tests were carried out by two different chromatographic methods of detection (PDA and ECD) in HPLC for quantify the phenolic compounds content and one methods (PDA) in HPLC for quantify the carotenoids content. The results of the study indicate that mainly the samples of leaves and flowers of *S. campanulata* have significant total phenolic content and the antioxidant activity independent of analyzed city. In relation to phytochemical tests, the leaves samples from Assis city presented the higher total content of phenolic compounds among analyzed cities independent of detector (PDA and ECD). Total content of carotenoids was higher the leaves samples from Londrina city. This study reveals that the flowers and leaves of *S. campanulata* present substances with antioxidant potential, as noted in the antioxidant tests and phytochemical screening. Thus, *Spathodea campanulata* can be considered a good source of antioxidants independent of the occurrence of the individual in different climatic regions of Brazil.

#### 1. Introduction

The Spathodea campanulata P. Beauv. (popularly named African tulip tree) belongs to the Bignoneaceae family, is native from African continent and was widely introduced in Brazil for ornamental purposes. Plants from Bignoneaceae family are known by producing compounds with aromatic, medicinal, and nutritional properties (Godzien et al., 2011; Manach et al., 2005). When these compounds are isolated, purified and incorporated in food supplements or medicaments, they can avoid metabolic deficiencies and prevent many kind of diseases. However, its leaves, flowers, and stem bark are used in folk medicine to treat edema, diarrhea, stomach complications, as well as, diuretic and anti-inflammatory (Choudhury et al., 2011; Heim et al., 2012; Ilodigwe et al., 2010). There has been increasing interest in bioactive compounds

produced by plants, especially when they have beneficial effects to human health, such as antioxidant activity. The antioxidant compounds, when isolated and identified, may be used for development and formulation of several products related to food, cosmetology and pharmaceuticals (Boligon et al., 2009; Zadra et al., 2012). Antioxidants are essentials for preventing the oxidative damage in living cells and oxidative deterioration in foods. These compounds can delay oxidative degradation reactions by complexing to pro-oxidant metals or reduce cellular structures oxidation by inhibiting the production or scavenging free radical (Anissi et al., 2014; Pietta, 2000).

Previous studies have demonstrated the presence of phenolic acids and flavonoids as the main phytochemicals in *S campanulata* aerial parts (Elusiyan et al., 2011; Nazif, 2007). However, as far as we know, there are no reports concerning antioxidant capacity, phenolic acids,

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flavonoids, and carotenoids, from leaves, flowers, and nectar of *S. campanulata*. Furthermore, plants from different climatic regions (two subtropical and one tropical area/cities) and two different chromato-graphic methods of detection were evaluated the possible climatic and analytical influences on the real content of phytochemicals.

# 2. Methods

#### 2.1. Chemicals and reagents

Analytical grade standards and organic solvents were obtained from Sigma Co. (St. Louis, MO, USA). All other chemicals and reagents were of analytical grade and purchased from J.T. Baker (JT Baker, Mexico).

#### 2.2. Plant material and climate zones

The leaves, flowers, and nectars were collected from specimens of *S. campanulata*, randomly selected, during 2016 year, and located in the urban perimeter of three different cities: Limeira ( $22^{\circ}34'34.0''S$ ,  $47^{\circ}21'54.1''W$ , climate - tropical), Assis ( $22^{\circ}38'53.2''S$ ,  $50^{\circ}25'12.8''W$ , climate - transition zone tropical/subtropical), and Londrina ( $23^{\circ}00'57.3''S$ ,  $51^{\circ}11'28.4''W$ , climate - subtropical) (Fig. 1). Leaves and flowers were collected and then dried in a forced air incubator at a temperature of 40 °C until reach constant weight. Dried samples were ground in a knife mill and the resulting powder was stored in amber flasks at -80 °C. The nectar was collected from completely opened flowers, and later freeze-dried and stored at -80 °C.

# 2.3. Extract preparation

Two-hundred milligrams of dried leaves, flowers, and nectar were homogenized in vortex for 30 s with 2 mL of HCl:H<sub>2</sub>O: methanol (2:18:80, v/v/v) and incubated for 30 min in an ultrasonic bath (Branson, Georgia, USA). After centrifugation for 10 min at 4 °C, the supernatant was collected and stored in amber flask. The pellet was reextracted thrice and the supernatants combined to obtain the methanolic extract (MEx). These extracts were filtered through 0,22  $\mu$ m syringe filter and used for antioxidant tests, total phenols quantification, and phenolic and flavonoids HPLC (UV/VIS -PDA detection and electrochemical detection - ECD) analysis. For carotenoids analysis in HPLC, 200 mg of powdered and dried samples were extracted with methanol and tetrahydrofuran (MeOH:THF extract) (Minatel et al., 2014). Antioxidant assays were performed in leaves and flowers MEx diluted 1:20 (v/v) in methanol. MEx of nectars were diluted 1:10 (v/v) in methanol only for ORAC assay.

# 2.4. Antioxidant assays

# 2.4.1. ORAC (Oxygen Radical Absorbance Capacity)

The ORAC values were determined using AAPH - 2,2'-Azobis (2methylpropionamidine) dihydrochloride to generate the peroxyl radical, as previously described (Bolling et al., 2012). Samples were incubated with 0.32 M AAPH and 10  $\mu$ M of fluorescein for 70 min. Fluorescence was monitored at 485 nm excitation wavelength and 520 nm emission using FLUOstar OPTIMA spectrofluorimeter (BMG LAB-TECH Inc., Cary, N.C., USA). A standard curve constructed in the range of 5 to 50  $\mu$ molTrolox/L was used to determine the ORAC values, based on the area under the response curve of each sample. The standard curve was linear between 0 and 50  $\mu$ M Trolox and the results are expressed as  $\mu$ M/TEAC.mg<sup>-1</sup> by dry weight of leaves, flowers and nectar.

# 2.4.2. FRAP (Ferric Reducing Antioxidant Power)

The FRAP assay was done according to Benzie and Strain (1996) with some modifications. Flowers and leaves methanolic extracts were diluted (1:20, v/v) in methanol. No dilution were made in nectars extracts. Samples were incubated in the dark at room temperature with the FRAP reagent for 1 h, and the absorbance at 593 nm was then recorded. FRAP values were calculated from standard curves using Trolox at 31.25–1000  $\mu$ M/L. Results were expressed as uM Trolox/mg by dry weight of leaves, flowers and nectar.

# 2.4.3. DPPH (2, 2-diphenyl-1-picrylhydrazyl) radical scavenging assay

The antioxidant activity of the leaves, flowers, and nectars were evaluated by monitoring their ability in quenching the stable free radical DPPH (Brand-Williams et al., 1995). Briefly, 900  $\mu$ L of 100  $\mu$ mol/L DPPH in ethanol was mixed with 100  $\mu$ L of sample and the decrease in absorbance was monitored at 517 nm until a constant reading was obtained. The % inhibition was calculated by:

$$\% inhibition = \frac{AB - AS}{AB} \times 100$$

where, AB = blank absorbance and AS = sample absorbance.

2.5. Total phenols

Total phenol of extracts were determined according to Singleton

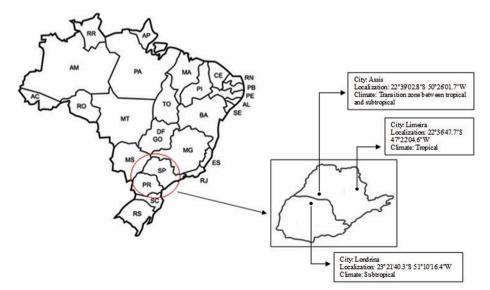


Fig. 1. Brazilian map showing the location of the study and the cities location.

et al. (1999). Two-hundred microliters of 10% Folin-Ciocalteu solution and 800  $\mu$ L of Na<sub>2</sub>CO<sub>3</sub> 700 mM were mixed to 100  $\mu$ L of each sample and incubated in the dark for 1 h at room temperature. Results were measured at 765 nm and expressed as micromol per liter gallic acid equivalents (GAE). The gallic acid standard curve ranged from 0.0125 to 0.2 mg GAE/mL.

# 2.6. HPLC/PDA analysis of phenols and flavonoids

The HPLC analysis of leaves, flowers, and nectar MEx were performed on a Thermo-Dionex UltiMate 3000 HPLC system (Thermo, Germany) equipped with quaternary pump, autosampler, column compartment, photodiode array detector and software Chromeleon version 7.1. A reversed phase ACE5 C18 (4.6 mm  $\times$  250 mm) column was used in the experiment. The injection volume was 10 µL, the column temperature maintained at 30 °C and the detection was set to 225, 280, 300, and 340 nm. The mobile phases used were: (A) phosphoric acid/ ultrapure water (1:99, v/v) and (B) methanol 70%, and the gradient was set up for 1.0 mL/min to begin at 100% solvent A followed by 70% solvent A over a 10-min linear gradient. This is followed by an 40-min linear gradient to 5% solvent A and 3-min hold at 5% solvent A, and finally, a 2-min linear gradient back to 100% solvent A. The system is held at 100% solvent A for 3 min for equilibration back to initial conditions. The compounds were identified by chromatographic comparisons with authentic standards.

# 2.7. HPLC/ECD analysis of phenolic acids and flavonoids

Phenolic acids and flavonoids were determined by HPLC with electrochemical detection (ECD), using equipment and conditions previously described (Chen et al., 2005). MEx of leaves and flowers were dissolved in the aqueous mobile phase at 1:10 (v/v) and 1:4 (v/v), respectively. The nectar MEx was not diluted.

# 2.8. HPLC/PDA analysis of carotenoids

Ten milliliters of MeOH:THF extracts, for each sample, were dried under nitrogen, resuspended in 100  $\mu L$  of ethanol and injected onto HPLC (equipment Thermo described above) equipped with a C30 column (3  $\mu m$ , 150  $\times$  3.0 mm; YMC, EUA), to determine the carotenoids content. The mobile phases A and B, running conditions, injection volume and flow rate were identical to previously described (Minatel et al., 2014).

### 2.9. Statistical analysis

All extractions and analysis were performed in triplicate. The results are expressed as means  $\pm$  standard deviations, and were analyzed by one-way analysis of variance (ANOVA) followed by the Tukey's test to multiple comparisons, using the SPSS software, version 19.0 (IBM Inc).

# 3. Results

# 3.1. Antioxidant capacity and total phenols

The antioxidant capacity and total phenols of leaves, flowers and nectar from *S. campanulata* growing in Limeira, Assis, and Londrina are showed in Table 1. The highest antioxidant capacity of each plant material were present in the climatic transition zone. In fact, for all climatic zones evaluated the antioxidant capacity and total phenols were higher in leaves, flowers and nectars, respectively.

Probably the temperature and rain incidence in transition zone is more favorable to antioxidant compounds synthesis by *S. campanulata* as showed by antioxidant assays and total phenols analysis. The ORAC assay was the most efficient procedure to assess the antioxidant capacity of the samples. This assay is more efficient to quantify the power of

#### Table 1

Antioxidant capacity and	total phenols	of Spathodea	campanulata.

Samples	Assis (transition)							
	DPPH (% inhibition)	FRAP <sup>a</sup> (uM TEAC/mg)	ORAC <sup>a</sup> (uM TEAC/mg)	Total phenols <sup>b</sup> (mg/g GAE)				
Leaves	$\textbf{3.98} \pm \textbf{0.002a}$	$\begin{array}{c} 16.40 \pm \\ 0.001a \end{array}$	$44.43 \pm 0.005a$	$\textbf{6.28} \pm \textbf{0.002a}$				
Flower	$\textbf{2.40} \pm \textbf{0.002b}$	10.17 ± 0.001b	$\begin{array}{c} 30.07 \pm \\ 0.004 \mathrm{b} \end{array}$	$\textbf{5.77} \pm \textbf{0.001b}$				
Nectar	$0.08\pm0.001c$	$\begin{array}{c} \textbf{0.281} \pm \\ \textbf{0.004c} \end{array}$	1.086 ± 0.007c	$0.25\pm0.003c$				
Londrina	(subtropical)							
Leaves	$\textbf{3.74} \pm \textbf{0.002a}$	$14.71 \pm 0.003a$	$\begin{array}{c} 40.37 \ \pm \\ 0.008^{\rm a} \end{array}$	$5.82\pm0.002a$				
Flower	$2.06\pm0.003b$	8.477 ± 0.002b	27.54 ± 0.005b	$5.21\pm0.002b$				
Nectar	$0.07\pm0.001c$	$\begin{array}{c} \textbf{0.243} \pm \\ \textbf{0.002c} \end{array}$	$0.956 \pm 0.002c$	$0.24\pm0.003c$				
Limeira (	tropical)							
Leaves	$3.54\pm0.003a$	$\begin{array}{c} 13.71 \pm \\ 0.002a \end{array}$	32.26 ± 0.003a	$5.35\pm0.003a$				
Flower	$1.80\pm0.001b$	$6.708 \pm 0.002b$	$23.29 \pm 0.005b$	$\textbf{4.62} \pm \textbf{0.002b}$				
Nectar	$0.05\pm0.001c$	$\begin{array}{c} 0.178 \ \pm \\ 0.001c \end{array}$	0.727 ± 0.001c	$0.22\pm0.002c$				

Values are expressed as mean  $\pm$  standard deviation; means with the same letter in the column do not differ by Tukey test ( $\alpha \le 0.05$ ).

 $^{\rm a}$  The antioxidant capacity is expressed in uM Trolox (TEAC)/mg of samples.  $^{\rm b}$  The total phenols are expressed in mg of samples per g of gallic acid equivalent.

radical scavenging in samples, once it has the ability to react with hydrophilic and hydrophobic compounds in a high extend than DPPH and FRAP assays.

# 3.2. Phenolic acids and flavonoids analysis by HPLC with PDA or ECD detector

In order to have a more efficient approach regarding the differences in phenolic acids and flavonoids profile, samples were evaluated by HPLC using two different detection systems. Employing both detection methods, PDA or ECD, the highest amounts of phenolic acids and flavonoids were found in leaves, flowers, and nectar, respectively. Despite the detection system employed and climatic zone, the leaves were the plant parts with the higher amounts of phenolic compounds, followed by flowers and nectar (Tables 2 and 3).

The PDA detection methods allowed the quantification of 5 compounds in leaves, 8 in flowers and 3 in nectar, independently of city analyzed. In relation to leaves, the most abundant compounds were caffeic acid and rutin, in flowers were caffeic acid and chlorogenic acid and in the nectar were identified only caffeic acid, p-cumaric and vitexin (Table 2).

Quantification and identification of phenolic compounds by HPLC/ ECD allowed to observed the presence of four compounds for leaves and flowers and three for nectar, independent of city. In leaves was observed the higher concentration of rutin and ferulic acid, for flowers' samples the most abundant compounds were chlorogenic and ferulic acid. The nectar's samples presented lower number of identified compounds, being the p-cumaric acid the major compound (Table 3).

Taking into account carotenoids analysis, it was observed the presence of six compounds for leaves and flowers samples and two compounds for nectar samples, for the cities Assis and Londrina. In the case of Limeira samples, the leaf presented three compounds, the flower showed two compounds and in the nectar sample does not possible identify none compound. Independently of analyzed city, the compounds the most abundant for leaves samples was the lutein (Assis -178,9  $\mu$ g g<sup>-1</sup>; Londrina - 378,2  $\mu$ g g<sup>-1</sup>; Limeira - 186  $\mu$ g g<sup>-1</sup>), flowers

#### Table 2

Phenolic acids and flavonoids identified by HPLC/PDA in leaves, flowers, and nectars of Spathodea campanulata.

Phenolic compound	ls (µg∕g dw)										Total
City (climate)	Samples	Caffeic acid	Chlorogenic acid	Ferulic acid	p-Cumaric acid	Gallic acid	Rutin	Isoorientin	Vitexin	Orientin	
Assis (transition)	Leaf	$\begin{array}{c} 1127.8 \pm \\ 0.18a \end{array}$	ND	46.1 ± 0.03b	ND	ND	$1127.0 \pm 0.01a$	ND	31.8 ± 0.03c	$\begin{array}{c} 0.3 \pm \\ 0.09 d \end{array}$	2333
	Flower	$571.4 \pm 0.01a$	$\textbf{857.1} \pm \textbf{0.04b}$	$\begin{array}{c} 55.1 \pm \\ 0.08 \mathrm{c} \end{array}$	$535.6 \pm 0.05d$	39.3 ± 0.02e	ND	0.86 ± 0.12f	$\begin{array}{c} 0.85 \pm \\ 0.05 f \end{array}$	$1.3~\pm$ 0.03g	2061
	Nectar	$0.097 \pm 0.11a$	ND	ND	$0.5\pm0.01b$	ND	ND	ND	$\begin{array}{c} \textbf{0.01} \pm \\ \textbf{0.01c} \end{array}$	ND	0.61
Londrina (subtropical)	Leaf	$961.9 \pm 0.05a$	ND	40.1 ± 0.07b	ND	ND	$\begin{array}{c} 1084.9 \pm \\ 0.04c \end{array}$	ND	$\begin{array}{c} \textbf{27.1} \pm \\ \textbf{0.02d} \end{array}$	$\begin{array}{c} \textbf{0.2} \pm \\ \textbf{0.04e} \end{array}$	2114
· • ·	Flower	$532.3 \pm 0.01a$	$696.4\pm0.09b$	$\begin{array}{c} 33.0 \pm \\ 0.05c \end{array}$	$\begin{array}{c} 488.0 \pm \\ 0.08d \end{array}$	$\begin{array}{c} 21.7 \pm \\ 0.03e \end{array}$	ND	$\begin{array}{c} \text{4.55} \pm \\ \text{0.07f} \end{array}$	$\begin{array}{c} \textbf{0.73} \pm \\ \textbf{0.02g} \end{array}$	$1.3 \pm 0.02 \mathrm{h}$	1973
	Nectar	$0.608 \pm 0.06a$	ND	ND	$2.3\pm0.10\text{b}$	ND	ND	ND	$\begin{array}{c} 0.02 \pm \\ 0.03c \end{array}$	ND	2.93
Limeira (tropical)	Leaf	$750.1 \pm 0.02a$	ND	40.7 ± 0.09b	ND	ND	$1045.1 \pm 0.03c$	ND	$\begin{array}{c} \text{27.5} \pm \\ \text{0.05d} \end{array}$	0.2 ± 0.11e	1864
	Flower	336.0 ± 0.11a	$406.2\pm0.11b$	$\begin{array}{c} 33.0 \pm \\ 0.04 c \end{array}$	$\begin{array}{c} 305.0 \pm \\ 0.02d \end{array}$	$\begin{array}{c} 14.7 \pm \\ 0.09e \end{array}$	ND	$\begin{array}{c} 0.17 \pm \\ 0.01 \mathrm{f} \end{array}$	$\begin{array}{c} 0.13 \pm \\ 0.03 f \end{array}$	$\begin{array}{c} 0.4 \pm \\ 0.07 \mathrm{f} \end{array}$	1096
	Nectar	$0.327 \pm 0.04a$	ND	ND	$2.3\pm0.08\text{b}$	ND	ND	ND	$\begin{array}{c} 0.01 \pm \\ 0.01c \end{array}$	ND	2.64

ND - Not detected.

Values are expressed as mean  $\pm$  standard deviation; means with the same letter in the line do not differ by Tukey test ( $\alpha \le 0.05$ ).

#### Table 3

Phenolic acids and flavonoids identified by HPLC/ECD in leaves, flowers and nectars of Spathodea campanulata.

Phenolic compounds	(µg/g dw)								
City (climate)	Samples	Gallic acid	Chlorogenic acid	Caffeic acid	p-Coumaric acid	Ferulic acid	Rutin	Quercetin 3- glucoside	Total
Assis (transition)	Leaf	ND	ND	$68.4 \pm \mathbf{0.03a}$	ND	$1415\pm0.01b$	238.4 ± 0.03c	$38.46 \pm \mathbf{0.05d}$	1760
	Flower	ND	$\textbf{466.9} \pm \textbf{0.02a}$	$13.1\pm0.03\text{b}$	$350.7\pm0.01c$	$\begin{array}{c} 410.4 \pm \\ 0.01 d \end{array}$	ND	ND	1241
	Nectar	$0.026 \pm 0.01a$	ND	$0.11\pm0.01b$	$1.4\pm0.02c$	ND	ND	ND	1.5
Londrina (subtropical)	Leaf	ND	ND	$\begin{array}{c} 61.03 \pm \\ 0.03a \end{array}$	ND	$1351\pm0.01b$	$\begin{array}{c} 176.8 \pm \\ 0.01c \end{array}$	$\textbf{35.92} \pm \textbf{0.03d}$	1624
	Flower	ND	$\textbf{438.4} \pm \textbf{0.04a}$	$\begin{array}{c} 13.85 \pm \\ 0.04b \end{array}$	$353.8\pm0.02c$	404.6 ± 0.02d	ND	ND	1210
	Nectar	$\begin{array}{c} 0.020 \ \pm \\ 0.03a \end{array}$	ND	$\begin{array}{c} 0.093 \pm \\ 0.02b \end{array}$	$0.3\pm0.02c$	ND	ND	ND	0.4
Limeira (tropical)	Leaf	ND	ND	$\begin{array}{c} 54.37 \pm \\ 0.07a \end{array}$	ND	$1299\pm0.03b$	$\begin{array}{c} 143.5 \pm \\ 0.01c \end{array}$	$40.99\pm0.02d$	1497
	Flower	ND	$233.9\pm0.02a$	$\begin{array}{c} 14.65 \pm \\ 0.02b \end{array}$	$128.4\pm0.02c$	327.9 ± 0.02d	ND	ND	705
	Nectar	$0.006 \pm 0.01a$	ND	$\begin{array}{c} 0.051 \pm \\ 0.03b \end{array}$	$1.4\pm0.06c$	ND	ND	ND	1.5

ND - Not detected.

Values are expressed as mean  $\pm$  standard deviation; means with the same letter in the line do not differ by Tukey test ( $\alpha \leq 0.05$ ).

samples was the *trans*-beta-carotene (Assis - 32,32  $\mu$ g g<sup>-1</sup>; Londrina - 30,82  $\mu$ g g<sup>-1</sup>; Limeira - 4,29  $\mu$ g g<sup>-1</sup>). In relation to nectars samples, the compound the most abundant was the *trans*-luteín (Table 4).

# 4. Discussion

The present study is the first to assess the bioactive compounds in leaves, flowers, and nectar of *S. campanulata* comparing different methods of analysis and climate zones. The maximum distance between cities was 484 km for Limeira and Londrina. In addition, the maximum and minimum temperature registered in 2016 for Assis, Londrina and Limeira cities were 29.3–9.7 °C; 29.6–10.7 °C, and 28.4–9.9 °C, respectively. Although the three cities showed temperatures relatively similar in the 2016 year, the phenolic compounds analyzed and the antioxidant activity from obtained extracts were always higher in plants

growing in Assis city. Based in these results, we hypothesized that phenolic compounds in *S. campanulata* are mainly synthesized when the trees are located in transition areas between warm (Limeira) and cold (Londrina) climates.

To measure, *in vitro*, the antioxidant capacity of a plant extract several laboratory methods are employed, such as DPPH, ORAC, FRAP, ABTS, and TBARS (Arnao et al., 1999; Pisoschi et al., 2016; Thaipong et al., 2006; Drouet et al., 2018). The evaluation of antioxidant activity is a complex process, because normally it can be occasioned for many mechanisms, therefore, it is advisable the utilization of more than one test. In this study, the DPPH, FRAP and ORAC tests, with samples of *S. campanulata*, showed significant antioxidant activity in all tests, mainly to flowers and leaves, giving emphasis to Assis, that showed the best results.

The test of total phenols provide great evidences to support the idea

#### Table 4

Carotenoids (µg/g d	w)									
City (climate)	Samples	Lutein	<i>Trans-</i> lutein	Zeaxanthin	Cryptoxanthin	13cis β-carotene	Alpha- carotene	<i>trans</i> -beta- carotene	9-cis-beta- carotene	Total
Assis (transition)	Leaf	$\begin{array}{c} 179.0 \pm \\ 0.03 a \end{array}$	ND	$5.3\pm0.04b$	ND	$\textbf{4.5} \pm \textbf{0.05b}$	$13.3\pm0.05c$	$62.7 \pm \mathbf{0.04d}$	$8.6\pm0.01e$	273.4
	Flower Nectar	$1.4 \pm 0.02a$ ND	ND 0.06 ± 0.02a	ND ND	$\begin{array}{c} 8.5 \pm 0.01 b\\ ND \end{array}$	$\begin{array}{c} 11.3 \pm 0.03 c\\ \text{ND} \end{array}$	$\begin{array}{l} \text{4.4} \pm 0.01 \text{d} \\ \text{ND} \end{array}$	$\begin{array}{c} 32.3 \pm 0.02e \\ 0.05 \pm 0.01a \end{array}$	$\begin{array}{l} \text{4.7} \pm 0.05 \text{d} \\ \text{ND} \end{array}$	62.6 0.1
Londrina (subtropical)	Leaf	$\begin{array}{c} 378.2 \pm \\ 0.01a \end{array}$	ND	$\begin{array}{c} 15.9 \pm \\ 0.05b \end{array}$	ND	$8.8\pm0.01c$	$\textbf{27.4} \pm \textbf{0.08d}$	$114.3\pm0.03e$	$16.1\pm0.01b$	560.7
	Flower Nectar	$\begin{array}{c} 2.3 \pm 0.01 a \\ \text{ND} \end{array}$	ND 0.05 ± 0.08a	$2.7\pm0.02a$ ND	$\begin{array}{c} 8.3 \pm 0.01 b\\ \text{ND} \end{array}$	$\begin{array}{l} \text{5.4} \pm 0.07c\\ \text{ND} \end{array}$	$\begin{array}{l} 4.6 \pm 0.02c \\ \text{ND} \end{array}$	$\begin{array}{c} 30.8 \pm 0.02d \\ 0.03 \pm 0.02a \end{array}$	ND ND	54.1 0.1
Limeira (tropical)	Leaf	$\begin{array}{c} 186.0 \pm \\ 0.05 \mathrm{a} \end{array}$	ND	$2.8\pm0.03b$	ND	ND	ND	$7.1\pm0.09c$	ND	155.9
	Flower Nectar	$3.2\pm0.03a$ ND	ND ND	ND ND	ND ND	ND ND	ND ND	$\begin{array}{c} 4.3 \pm 0.01 b\\ ND \end{array}$	ND ND	7.5 -

ND - Not detected.

Values are expressed as mean  $\pm$  standard deviation; means with the same letter in the line do not differ by Tukey test ( $\alpha \leq 0.05$ ).

that antioxidant activity present in the *S.campanulata* is linked to the presence of polyphenols, since that the aim of test is measure the content of phenolic compounds in the evaluated sample (Pisoschi et al., 2016). In this study, the total phenols presented significant results for flowers and leaves samples, mainly to city of Assis, like noted previously in the antioxidant tests.

*Spathodea campanulata* becomes one interesting specie of study, once previous researches has demonstrated potential antioxidant activity from some parts of this plant (Heim et al., 2012; Sangeetha et al., 2016). Nevertheless, few information regarding the phytochemical associated to this antioxidant activity are available. In the present study, we identified and quantified phenolic compounds by applying two different liquid chromatography detection systems (PDA and ECD). Each detector can present more or less capacity to detect certain substances, depending on the chemical origin of compound (Williamson et al., 2003; Elwekeel et al., 2013; Chambers et al., 2017). Both detectors were efficient, but compounds as trans ferulic acid, isoorientin, vitexin and orientin were identified only by PDA detector. On the other hand, ferulic acid and quercetin 3-glucoside were identified only by ECD. The other metabolites present in Tables 2 and 3 differed with regard their concentrations.

Despite of electrochemical (EC) detector is considered one of the most sensitive and selective HPLC detectors available and the PDA detector is considered more generalist (Kissinger and Heineman, 1984; Ackworth, 1997), in this study the photodiode (PDA) detector was more sensitive, in the number of compounds as in the concentration of identified compounds. EC detectors require the use of electrically conductive mobile phases capable of oxidize and reduce the compounds present into the analyzed samples, by any chance the reduction or oxidation do not happen, others detector, such as PDA, it is become more efficient and sensitive than ECD. In the present study, maybe the mobile phase used in the analysis with EC detector was not capable of oxidize and reduce the phenolic compounds present in samples, once these compounds are present in many oxidation and reduction reactions. Due the limitations and particularities of both detectors, some studies of chemical investigation of complex samples such as plant samples, it advises the use EC and PDA detectors, in order to maximize the identification of the compounds in the sample (Gamborg et al., 1961; Pereira et al., 2009).

The synthesis of secondary compounds in general suffers the direct influence of intrinsic factors of ambient, for example, solar radiation, photoperiod, temperature, soil types, rainfall and even biotic factors as action of herbivorous, in this way, plants of same species can present greater or lesser capacity of synthesis of bioactives depending on where it is inserted (Verpoorte et al., 1999; Inderjit and Duke, 2003; Carrier et al., 2003; Khan et al., 2009). A study carried out in leaves of 93 plants,

located in two African rain forests, showed a remarkable difference in phenolic content between forests (Gartlan et al., 1980). The authors attributed this difference to soil nutrients and the region climate.

Plants with low levels of Ni can present decreased production of anthocyanins (Hawrylak et al., 2007). Trace metals obviously limit anthocyanin biosynthesis by inhibiting activity of L-phenylalanine ammonia-lyase (PAL) (Krupa et al., 1996). In studies performed by Angelova (et al., 2006) the nutrients AgNO<sub>3</sub> and CdCl<sub>2</sub> elicited overproduction of two tropane alkaloids, scopolamine and hyoscyamine, by in hairy root cultures of Brugmansia candida. According to Chambers et al. (2017), soils with higher quantities of macro elements, such as calcium and magnesium, can impact on the content of crude fiber and in the concentration of flavonolignans in Silybum marianum. Other abiotic factor, temperature variations, it has multiple effects on the metabolic regulation, permeability, rate of intracellular reactions in plant cell cultures, likewise, it may influence in the secondary metabolite production of plants, as reported in studies performed with Melastoma malabathricum cell cultures incubated at a lower temperature range (20  $\pm$  2 °C) grew better and had higher anthocyanin production than those grown at 26  $\pm$  2 °C and 29  $\pm$  2 °C (Chan et al., 2010) and studies carry out by Szakiel et al. (2011), authors noted that lower soil temperatures caused an increase in levels of steroidal furostanol and spirostanol saponins.

Despite of some antioxidant tests as ORAC and Folin-Ciocalteu method have a higher sensibility in the detection of hydrophilic compounds, these methods also can interact with others compounds of lipophilic origin, such as carotenoids, terpenes and some classes of vitamins, being necessary more sensitive tests as high-performance liquid chromatography (HPLC) to identification of bioactives present in the test samples (Antolovich et al., 2002; De Oliveira et al., 2009).

Phytochemical investigations performed with species of Bignoneaceae family showed the presence of many classes of secondary metabolites, which goes from tannins, flavonoids, phenolic acids until lipophilic compounds as carotenoids (De Oliveira et al., 2009; Antonisamy et al., 2012). In relation to investigation carried out in this study, the phenolic compounds were identified and quantified by two detectors, the photodiode (PDA) and electrochemical (ECD), because each detector can present more or less capacity of detect of certain substances, depending on the chemical origin of compound (Williamson et al., 2003). Both detectors were efficient, but compounds as trans ferulic acid, isoorientin, vitexin and orientin were identified only by PDA detector, ferulic acid and quercetin 3-glucoside were identified only by ECD. The other metabolites present in Tables 2 and 3 differed with regard their concentrations. Despite of electrochemical (EC) detector is considered one of the most sensitive and selective HPLC detectors available (Kissinger and Heineman, 1984; Ackworth, 1997), in this study the photodiode (PDA) detector presented higher effectiveness, in either in the number of identified compounds and concentration of identified compounds. EC detectors require the use of electrically conductive mobile phases capable of oxidize and reduce the compounds present into the analyzed samples, by any chance the reduction or oxidation do not happen, others detector, such as PDA, it is become more efficient and sensitive than ECD (Michael, 2010). In the present study, maybe the mobile phase used in the analyze with EC detector was not capable of oxidize and reduce the phenolic compounds present in samples, once these compounds are present in many oxidation and reduction reactions (Gamborg et al., 1961; Pereira et al., 2009).

Among the identified polyphenols by both detectors, we can highlight the rutin, caffeic acid, ferulic acid, compounds noted in previous studies carried out with the *S. campanulata* and known in scientific literature as antioxidant substances (Kowti et al., 2011; Heim et al., 2012; Coolborn et al., 2015). Ilodigwe et al. (2010) showed that leaves of species present polyphenols in its chemical composition, such as phenolic acids (caffeic acid, ferulic acid, gallic acid) and some classes of flavonoids (kaempferol, quercetin, quercetin 3-glucoside and rutin). In the case of screening performed with flowers, it was possible identify others classes of polyphenols as tannins and some flavonoids, as quercetin and rutin (Banerjee and DE, 2001).

In relation to therapeutic potential of polyphenols, studies performed with rutin showed that the compound has capacity of acts by inhibiting the formation of free radicals in many levels, because the flavonoid can react with anion superoxide and lipid peroxyl radicals neutralizing them, and it can also form a complex with iron that catalyze the formation of radical of non-toxic oxygen (Kim et al., 2011; Radjabian and Huseini, 2010; Poppe and Petersen, 2016). The caffeic acid, other class of phenolic compounds, also present many therapeutic properties with highlight for its antioxidant activity (Heim et al., 2012). Nardini et al. (1995) showed that the caffeic acid inhibited, in a dose/dependent manner, the lipid peroxidation induced by cupric ions. In the same study, the authors showed that the phenolic acid is capable of reduction the lipoperoxyl radicals (ROO•) (by means of donation of a hydrogen) and imped the propagation of lipid peroxidation. Finally, the ferulic acid, it has an antioxidant action due its chemical structure that facilitate the neutralization of free radicals, mainly the hydrogen peroxide, superoxide anions and hydroxyl radicals, reducing the degradation of proteins and lipids, caused mainly by hydroxyl radicals. Moreover, the ferulic acid also showed effective in the regulation and renovation of enzymes involved in the combat to the oxidative stress, such as superoxide dismutase (SOD) and catalase (Coolborn et al., 2015)).

Others compounds with therapeutic potential that were identified in this study were the carotenoids (Table 4). The results of analyzed samples of cities (Assis, Londrina e Limeira) showed the presence mainly of lutein and *trans*-beta-carotene, substances known for their medicinal potential. There are no studies reporting the presence of carotenoids in *S. campanulata*, Tinoi et al. (2006), showed the presence this class in *Pyrostegia venusta* and *Tabebuia chrysantha*, species of Bignoneaceae family, and the majority compounds were the lutein, zeaxanthin, cryptoxanthin and  $\beta$ -carotene. Foods rich in carotenoids as lutein, zeaxanthin and  $\beta$ -carotene may help ophthalmic damage prevention, mainly due their antioxidant activity, by scavenging free radical and working as filters for blue light and high ultraviolet energy (Rodriguez-Amaya et al., 2008).

Together, the antioxidant tests and phytochemical screening, can complement other potentiality assigned to *S. campanulata*, the insecticidal activity from nectar (Amusan et al., 1996; Santos et al., 2017). Researchers believe that the nectar of specie present compounds in its chemical composition with insecticidal activity and oxidant and does not compounds with antioxidant activity (Portugal-Araújo, 1963; Santos et al., 2017). This idea is supported, because in studies of count of dead insects inside the flowers of *S. campanulata* showed the presence of many species, as *Plebeia droryana, Tetragonisca angustula, Scaptotrigona postiça* and *Trigona spinipes* (Nogueira-Neto, 2002). Portugal-Araújo (1963) considered also the toxic effects of nectar from *S. campanulata*, where the researcher noted the presence of approximately 200 dead insect, among ants, bees and dipterous in just a single inflorescence this specie. Finally, Santos et al. (2017) showed the toxic effect of gross and dialyzed nectar in the insecticide tests with *E. heros, H. zea* and *A. gemmatalis*. The authors believe that insecticidal activity from nectar be caused indirectly by classes of identified proteins in study, as glycosyl transferase family and serine-threonine-protein phosphatase, since the proteins are involved in the biosynthesis of secondary compounds with insecticidal activity (Ferreyra et al., 2013).

In summary, this study reveals that the flowers and leaves of *S. campanulata* present substances with antioxidant potential, as noted in the antioxidant tests and phytochemical screening. Thus, *Spathodea campanulata* can be considered a good source of antioxidants independent of the occurrence of the individual in different climatic regions of Brazil.

#### Declaration of competing interest

The authors this article have no affiliations with or involvement in any organization or entity with any financial interest (such as honoraria; educational grants; participation in speakers' bureaus; membership, employment, consultancies, stock ownership, or other equity interest; and expert testimony or patent-licensing arrangements), or nonfinancial interest (such as personal or professional relationships, affiliations, knowledge or beliefs) in the subject matter or materials discussed in this manuscript.

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#### Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.bcab.2020.101536.

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